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## Tauroacidins A and B, New Bromopyrrole Alkaloids Possessing a Taurine Residue from *Hymeniacidon* Sponge

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Abstract: Two new bromopyrrole alkaloids, tauroacidins A (1) and B (2), with tyrosine kinase inhibitory activity have been isolated from an Okinawan marine sponge Hymeniacidon sp. and the structures were elucidated on the basis of spectral data and chemical means. Tauroacidins A (1) and B (2) are rare bromopyrrole alkaloids possessing a taurine residue. © 1997 Elsevier Science Ltd.

During our search for bioactive substances from marine organisms, we have investigated extracts of an Okinawan marine sponge *Hymeniacidon* sp. and isolated two new bromopyrrole alkaloids possessing a taurine residue, tauroacidins A (1) and B (2), with tyrosine kinase inhibitory activity. Here we describe the isolation and structure elucidation of 1 and 2.

*n*-BuOH-soluble materials of MeOH extract of the sponge collected off Ishigaki Island, Okinawa, were subjected to Sephadex LH-20 (MeOH) and C<sub>18</sub> columns (CH<sub>3</sub>CN/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H) followed by C<sub>18</sub> HPLC (MeOH/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H) to yield tauroacidins A (1, 8 x 10<sup>-5</sup> %, wet weight) and B (2, 5 x 10<sup>-5</sup> %).

HRESIMS [m/z 526.9164, (M-H)-,  $\Delta$  +0.9 mmu] of tauroacidin A {1, [ $\alpha$ ]D<sup>28</sup> -4.3° (c 0.15 MeOH)}, indicated the molecular formula to be C<sub>13</sub>H<sub>16</sub>N<sub>6</sub>O<sub>5</sub>SBr<sub>2</sub>. The IR spectrum of 1 suggested the presence of OH and/or NH (3420 cm<sup>-1</sup>), amide carbonyl (1690 cm<sup>-1</sup>), and sulfonate groups (1210 and 1040 cm<sup>-1</sup>).<sup>2</sup> The UV absorption [ $\lambda_{max}$  273 nm ( $\epsilon$  9700)] was attributable to a substituted pyrrole chromophore.<sup>3</sup> The <sup>1</sup>H and <sup>13</sup>C NMR data showed signals due to a 2,3-dibromopyrrole carbonyl moiety.<sup>4</sup> The <sup>13</sup>C chemical shifts at C-2' ( $\delta_{C}$  39.8) and C-3' ( $\delta_{C}$  49.1) corresponded well to those of the taurine residue (ca.  $\delta_{C}$  39 and 48, respectively) of melemeleone A.<sup>5</sup> The presence of the taurine residue in 1 was also supported by amino acid analysis of the acid hydrolysis products of 1. The whole structure of tauroacidin A (1) was elucidated by analyses of 2D NMR data [<sup>1</sup>H-<sup>1</sup>H COSY, NOESY, and decoupled HMBC (D-HMBC)<sup>6</sup>] (Fig. 1). The <sup>1</sup>H-<sup>1</sup>H COSY spectrum of 1 showed proton networks from NH-7 to H-10 and OH-9 and from NH-1' to H<sub>2</sub>-3'. The presence of the aminoimidazole ring including three quaternary carbons (C-11, C-13, and C-15) was revealed

	1		2	
positn.	$\delta_{\rm H}$ (m, $J^{a}$ )	$\delta_{C}$ (m)	$\delta_{H}$ (m, $J^{a}$ )	$\delta_{C}$ (m)
1.	12.70 (s)		11.84 (br.s)	
2.		104.7 (s)	6.99 (br.s)	121.3 (d)
3.		97.8 (s)		94.9 (s)
4.	6.95 (s)	113.0 (d)	6.87 (br.s)	111.8 (d)
5.		127.9 (s)		126.7 (s)
6.		159.1 (s)		159.9 (s)d
7.	8.22 (t, 6.0)		8.20 (t, 6.2)	
8.	3.32 (m) <sup>b,c</sup>	44.5 (t)	3.34 (m) <sup>b,c</sup>	44.3 (t)
9.	4.61 (m)	67.3 (d)	4.62 (m)	67.2 (d)
9-OH.	5.94 (d, 4.2)		5.95 (br.s)	
10.	6.17 (d, 4.2)	116.4 (d)	6.18 (m)	116.5 (d)
11.		131.2 (s)		131.1 (s)
12.	9.31 (br.s)		9.32 (br.s)	
13.		165.9 (s)		165.8 (s) <sup>c</sup>
14.	10.51 (s)		10.50 (s)	
15.		167.6 (s)		167.6 (s) <sup>c</sup>
16.	8.05 (br.s)		8.05 (br.s)	
1'	9.64 (br.t, 3.0)		9.65 (m)	
2'	3.67 (dt, 3.0, 7.2) <sup>c</sup>	39.8 (t)	3.67 (m) <sup>c</sup>	39.5 (t) <sup>e</sup>
31	2.75 (t, 7.2) <sup>c</sup>	49.1 (t)	2.74 (t, 7.5) <sup>c</sup>	49.2 (t)

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR Data of Tauroacidins A (1) and B (2) in DMSO-d<sub>6</sub>.

a) in Hz. b) These signals were overlapped with H<sub>2</sub>O signal. c) 2H. d) These signals were assigned on the basis of D-HMBC correlations. e) This signal was overlapped with DMSO signal.

by the D-HMBC cross-peaks for H-10/C-15, NH-14/C-13, and NH-14/C-15. D<sub>2</sub>O-Exchangeable protons at  $\delta_{\rm H}$  9.31 (NH-12) and 8.05 (NH-16) were assigned by NOESY correlations for H-9/NH-12 and NH-12/NH-16. Attachment of the pyrrole ring to N-7 via an amide bond was deduced from the NOESY cross-peak for H-4/NH-7 and D-HMBC correlations for H-4/C-6 and NH-7/C-6. The NOE for NH-1'/H-10 indicated Z-geometry of the double bond at C-10. The D-HMBC cross-peaks for NH-1'/C-11 and NH-1'/C-15 and the NOE for NH-1'/H-10 implied that the taurine residue was attached to C-15 of the imidazole ring. The negative ion FAB MS/MS spectrum of the pseudomolecular ion (m/z 527) of 1 showed several product ions supporting the proposed structure (Fig. 2). Thus the structure of tauroacidin A was concluded to be 1. In order to dertermine the absolute configuration at C-9, tauroacidin A (1) was subjected to ozonolysis and then hydrolysis.<sup>7</sup> Amino acid analysis of the acid hydrolysate showed the presence of one molar equivalent of isoserine. The hydrolysate was treated with HCl/i-PrOH and then (S)-MTPACI to afford the N,O-bis-(R)-MTPA/i-propyl ester of isoserine, of which HPLC analyses revealed the presence of (S)- and (R)-isoserines in the ratio of ca. 6:4, showing that tauroacidin A (1) was a 6:4 mixture of 9S- and 9R-isomers.

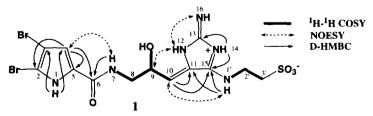


Fig. 1. 2D NMR Correlations for Tauroacidin A (1)

Fig. 2. Negative FABMS/MS Fragmentations of m/z 525 from Tauroacidins A (1)

The molecular formula,  $C_{13}H_{17}N_6O_5SBr$ , of tauroacidin B (2), optically inactive, was suggested by HRESIMS. The <sup>1</sup>H and <sup>13</sup>C NMR data of 2 differed from 1 only in the presence of an  $sp^2$  methine signal [H-2:  $\delta_H$  6.99 (m), C-2:  $\delta_C$  121.3 (d, <sup>1</sup> $J_{CH}$  = 185 Hz)], indicating the presence of 3-bromopyrrole carbonyl moiety<sup>4,8</sup> in 2. Thus the structure of tauroacidin B (2) was elucidated to be 2-debromo form of 1. Tauroacidin B (2) was revealed to be racemic (9S/9R = 1:1) at C-9 from HPLC analyses of the  $N_iO$ -bis-(R)-MTPA/i-propyl ester of isoserine contained in the hydrolysate of 2 obtained by the same procedure as 1.

Tauroacidins A (1) and B (2) are new bromopyrrole alkaloids possessing a taurine residue attached to the aminoimidazole ring, which may be biogenetically related to mauritamide A, known bromopyrrole alkaloids poessessing a taurine residue at C-11 from sponges.<sup>9</sup> Tauroacidins A (1) and B (2) exhibited inhibitory activity against EGF receptor kinase and *c-erbB*-2 kinase (IC<sub>50</sub>, 20 μg/mL each).

## **EXPERIMENTAL**

**General Procedure.** Optical rotations were recorded on a JASCO DIP-360 polarimeter. The IR and UV spectra were taken on a JASCO FT/IR-5300 and a JASCO Ubest-35 spectrophotometers, respectively. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AMX-600 and a JEOL EX-400 spectrometers, respectively. ESI mass spectra were obtained on a JEOL SX-102A spectrometer. Standard amino acid analyses were performed with Hitachi amino acid autoanalyzer Model 835.

**Sponge Materials.** The sponge *Hymeniacidon* sp. (order Halichondrida; family Halichondriidae) was collected off Ishigaki Island, Okinawa, and kept frozen until used. Soft and compressible sponge is somewhat fleshy texture. No ectosomal skeleton is just a thickening in this region of the mesohyl. Spicules are in loose tracts or without orientation. Spicules are styles 439 x 12 µm that are curved along the upperthird of the shaft. The voucher specimen (SS-361) was deposited at the Faculty of Pharmaceutical Sciences, Hokkaido University.

**Extraction and Isolation.** The sponge (5.5 kg, wet weight) was extracted with methanol (3 L x 2). Part (100 g) of the methanolic extract (335.8 g) was partitioned between ethyl acetate (500 mL x 3) and water, and the aqueous layer was extracted with *n*-butanol (1 L x 3). The *n*-butanol soluble material (18.6 g) was subjected to a Sephadex LH-20 column (MeOH), a C<sub>18</sub> column (Develosil ODS-LOP, Nomura Chemical, 45 x 490 mm; CH<sub>3</sub>CN/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H, 20:80:0.1), and C<sub>18</sub> HPLC (Develosil ODS-HG-5, Nomura Chemical, 10 x 250 mm; MeOH/H<sub>2</sub>O/CF<sub>3</sub>CO<sub>2</sub>H, 46:54:0.1; flow rate, 2.5 mL/min; UV detection at 260 nm) to yield tauroacidins A (1, 8 x 10<sup>-5</sup> % wet weight, t<sub>R</sub> 30 min) and B (2, 5 x 10<sup>-5</sup> %, t<sub>R</sub> 18 min).

**Tauroacidin A (1).** A colorless amorphous solid;  $[\alpha]_D^{28}$  -4.3° (c 0.15, MeOH); UV (MeOH)  $\lambda_{max}$  273 ( $\epsilon$  9700) and 312 nm (3800); IR (film)  $\nu_{max}$  3420 (br), 1690, 1630, 1210, and 1040 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (see Table 1); ESIMS (Neg., MeOH) m/z 525, 527, and 529 [(M-H)<sup>-</sup>, ca. 1:2:1]; HRESIMS m/z

526.9164 (M-H), calcd for C<sub>13</sub>H<sub>15</sub>N<sub>6</sub>O<sub>5</sub>S<sup>79</sup>Br<sup>81</sup>Br, 526.9173.

**Tauroacidin B (2).** A colorless amorphous solid;  $[\alpha]_D^{28}$  0° (c 0.10, MeOH); UV (MeOH)  $\lambda_{max}$  272 (ε 10000) and 310 nm (4000); IR (film)  $\nu_{max}$  3420 (br), 1690, 1630, 1210, and 1040 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (see Table 1); ESIMS (Neg., MeOH) m/z 447 and 449 [(M-H)<sup>-</sup>, ca. 1:1]; HRESIMS m/z 447.0098 (M-H)<sup>-</sup>, calcd for  $C_{13}H_{16}N_6O_5S^{79}Br$ , 447.0088.

Hydrolysis of Tauroacidins A (1). Tauroacidin A (1, 0.1 mg) was treated with 6 N HCl (100  $\mu$ L) at 110 °C for 24 h. After evaporation of the slovent, standard amino acid analysis of the reaction mixture showed the presence of 1 mole of taurine.

Absolute Stereochemistry at C-9 of Tauroacidins A (1) and B (2). A solution of tauroacidin A (1, 0.2 mg) in MeOH (100 μL) was treated with ozone at -78 °C for 2 min. After excess ozone was removed by N<sub>2</sub> gas, to the residue were added HCO<sub>2</sub>H (100 μL) and 35 % H<sub>2</sub>O<sub>2</sub> aq (100 μL). The mixture was stirred for 1 h at 0 °C and then 5 h at room temperature. After evaporation, the residue was hydrolyzed with 6 N HCl (100 μL) at 110 °C for 24 h in a sealed tube. Standard amino acid analysis of the hydrolysate showed the presence of 1 mole each of taurine and isoserine. The hydrolysate was treated with 9 % HCl/i-PrOH (100 μL) at 110 °C for 30 min, and then treated with CH<sub>2</sub>Cl<sub>2</sub> (100 μL), 4-(N,N-dimethylamino)pyridine (0.1 mg), Et<sub>3</sub>N (10 μL), and (S)-MTPACl (5 μL) at 40 °C for 2 h. After addition of N,N-dimethyl-propane-1,3-diamine (5 μL), the reaction mixture was subjected to HPLC analysis [YMC Pack SIL-06, 4.6 x 250 mm; flow rate, 1 mL/min; UV detection at 240 nm; eluent, hexane/i-PrOH, 95:5]. Retention times of N,O-bis-(R)-MTPA/i-propyl ester derivatives of authentic (S)- and (R)-isoserine were 13.4 and 16.3 min, respectively. Both (S)- and (R)-isoserine derivatives were found in the hydrolysate of 1 in the ratio of ca. 6:4. HPLC analysis of N,O-bis-(R)-MTPA-isoserine i-propyl ester derived from the hydrolysate of tauroacidin B (2) was performed under the same condition as described above, and both (S)- and (R)-isoserine derivatives were found in the ratio of ca. 1:1.

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## References and Notes

- 1. Kobayashi, J.; Tsuda, M.; Fuse, H.; Sasaki, T.; Mikami, Y. J. Nat. Prod., 1997, 60, 150-154.
- 2. Pretsch, E.; Clerc, T.; Seibl, J.; Simon, W. In Tables of Spectral Data for Structure Determination of Organic Compounds 2nd Edition; Biemann, K.; Springer-Verlag, Berlin, 1989; p1230 and p1162.
- 3. Scott, A. I. In *Interpretation of the Ultraviolet Spectra of Natural Products*; Pergamon Press: New York, 1964; p165-169.
- 4. Kobayashi, J.; Ohizumi, Y.; Nakamura, H.; Hirata, Y. Experientia, 1986, 42, 1176-1177.
- Alvi, K. A.; Diaz, M. C.; Crews, P.; Slate, D. L.; Lee, R. H.; Moretti, R. J. Org. Chem., 1992, 57, 6604-6608
- 6. Furihata, K.: Seto, H. Tetrahedron Lett., 1995, 36, 2817-2820.
- 7. Ozonolysis condition; Kobayashi, J.; Honma, K.; Sasaki, T.; Tsuda, M. Chem Pharm. Bull., 1995, 43, 403-407.
- 8. Kalinnowski, H.-O.; Berger, S.; Braun, S. In Carbon-13 NMR Spectroscopy; John Wiley & Sons: Chichester, 1988; p506.
- 9. Jiménez, C.; Crews, P. Tetrahedron Lett., 1994, 35, 1375-1378.